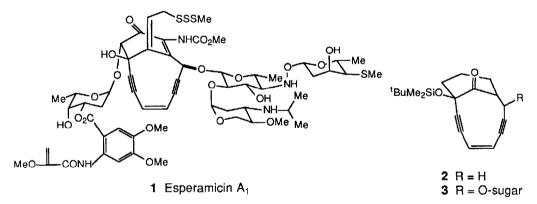
SYNTHESIS OF THE CORE ENEDIYNE STRUCTURE OF ESPERAMICIN-CALICHEMICIN CLASS OF ANTITUMOR ANTIBIOTICS

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Summary: The core structure of esperamicin-calichemicin series, bicyclo[7.3.1]enediyne 2, was synthesized in 9% overall yield in eight steps from cyclohexane-1,2-dione and was shown to be stable at room temperature.

The heart of esperamicin-calichemicin class of novel antitumor antibiotics (e.g., 1) is a bicyclo[7.3.1]tridecane system that combines a methyl trisulfide group, an endocyclic α , β -unsaturated ketone, and double and triple bonds in a 3-ene-1,5-diyne relationships.²⁻⁴ Enzymatic reduction of a trisulfide to a thiol triggers intramolecular Michael addition of the thiol to the enone, and subsequent Bergman enediyne cyclization to a phenylene diradical which can abstract protons from deoxyribose units in deoxyribonucleic acid resulting DNA strand breakage. Inspired by these fascinating structures, much efforts have been devoted toward design, synthesis, and evaluation of simple structures that might mimic the biological action of natural products.⁵



As a part of our continuing program aimed at the asymmetric total synthesis of antitumor natural products,⁶ we have also engaged in this attractive and challenging field and our initial goal for the synthetic studies is bicyclo[7.3.1]enediynes 2, 3 which are expected to be stable at the ambient temperature toward Bergman cyclization due to the presence of carbonyl group at the bridging chain. Quite recently the synthesis of compound 2 has been reported by Magnus and his co-workers.^{5b,f} In the present communication, we report an independent synthesis of a bicyclo[7.3.1]tridecane 2 which contains an enediyne unit in a requisite linear arrangement.

Our synthesis of 2 began with the reaction of cyclohexane-1,2-dione 4 with acetylene.⁷ Treatment of 4 with ethynylmagnesium bromide in THF at room temperature for 0.5 h afforded 5 as colorless needles (mp 50-51 °C) in 76% yield.⁸ Protection of 5 with *t*-butyldimethylsilyl group (*t*-BuMe₂SiOTf/2,6-lutidine/CH₂Cl₂, 0 °C, 1.5 h) provided 6 as colorless plates (mp 38-40 °C) in 98%

yield. Stephans-Castro coupling reaction⁹ of 6 with Z-dichloroethene $(Pd(PPh_3)4/CuI/i-Pr_2NH/benzene, rt, 1 h)$ provided 7 as a colorless oil in 79% yield. Further coupling reaction of 7 with methyl propargyl ether $(Pd(PPh_3)4/CuI/BuNH_2/benzene, rt, 12 h)$ afforded 8 as a colorless oil in 53% yield. Me_3Si-enol ether 9 and t-BuMe_2Si-enol ether 10 were prepared from 8 by treating with Me_3SiOTf and t-BuMe_2SiOTf in refluxing Et_3N in 86 and 88% yields, respectively.¹⁰

Treatment of Me₃Si-enol ether 9 and t-BuMe₂Si-enol ether 10 with Co₂(CO)₈ in *n*-heptane at room temperature for 2 h provided the corresponding dicobalt hexacarbonyl adducts, 11 and $12,^{11,12}$ in 68 and 88% yields, respectively.

Cyclization of 11 and 12 was carried out under the standard Nicholas conditions.^{11,12} Thus, treatment of Me₃Si-enol ether 11 with TiCl₄ (3.0 equiv)/DABCO (1.0 equiv) in CH₂Cl₂ at -60 °C for 10 min smoothly provided 13 as a brown solid in 64% yield. On the other hand, treatment of *t*-BuMe₂Si-enol ether 12 afforded 14 as a major product, probably arising from hydroxylation of the corresponding allylic carbocation species generated by desilyloxylation.¹³ We believe that trimethylsilyl group facilitates the attack of enol-double bond to terminal carbocation species, due to the less bulkiness of the group than *t*-butyldimethylsilyl group.

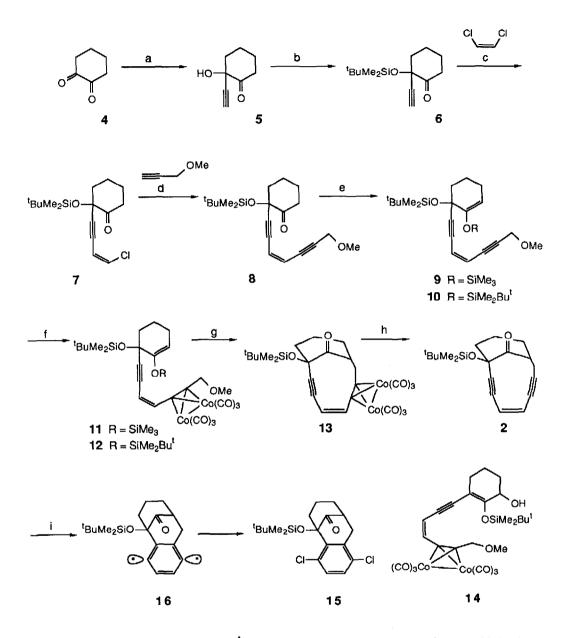
Decomplexation of **13** with Me₃NO (10 equiv) in MeOH at room temperature for 12 h provided the desired bicyclo[7.3.1]enediyne **2** as colorless needles (dp 42-45 °C) in 73% yield without formation of any detectable amount of Bergman reaction product.¹⁴

Bergman cyclization of 2 took place by heating at reflux for 4 h in CCl4 to afford dichloride 15 in 30% yield, indicating involvement of a phenylene diradical intermediate 16 in the reaction. Evaluation of biological activity of 2 is in progress and will be reported elsewhere.

Further studies toward the synthesis of esperamicin-calichemicin aglycone are in progress in our laboratories.

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a) HC=CMgBr/THF, rt, 0.5 h, 76%; b) ^tBuMe₂SiOTf/2,6-lutidine/CH₂Cl₂, 0 °C, 1.5 h, 98%; c) *Z*-dichloroethene/Pd(PPh₃)₄/Cul/*i*-Pr₂NH/benzene, rt, 1 h, 79%; d) methyl propargyl ether/ Pd(PPh₃)₄/Cul/BuNH₂/benzene, rt, 12 h, 53%; e) Me₃SiOTf/Et₃N, reflux, 1 h, 86% for **9** (^tBuMe₂SiOTf/Et₃N, reflux, 2.5 h, 88% for **10**); f) Co₂(CO)₈/*n*-heptane, rt, 2 h, 68% for **11** (88% for **12**); g) **11**-TiCl₄ (3.0 equiv)/DABCO (1.0 equiv)/CH₂Cl₂, -60 °C, 10 min, 64%; h) Me₃NO/MeOH, rt, 12 h, 73% i) reflux in CCl₄, 4 h, 30%

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- 13. It is important to note that Magnus^{5f} reported a successful Nicholas reaction of *t*-BuMe₂Si-enol ether 12 providing 13 as a major product. In our carefully controlled reaction, formation of only a trace amount of 13 was observed.

In a model reaction with use of Me₃Si-enol ether and *t*-BuMe₂Si-enol ether of cyclohexanone, Me₃Si-enol ether provided alkylation product (dicobalt hexacarbonyl adduct of methyl propargyl ether/TiCl₄/DABCO/CH₂Cl₂) in 73% yield, whereas *t*-BuMe₂Si-enol ether did not provided isolable amount of alkylation product.

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